

# SYNTHESIS OF 1-AMINO-2-ARYLETHYLENEIMINE ARYLHYDRAZONES

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In a continuation of our research [1] on the inversion of nitrogen in 1-aminoethylene-imine derivatives we synthesized 1-amino-2-arylethyleneimine arylhydrazones (Ia-d) via the reaction of dimethylsulfoniummethylide with symmetrical azines in order to accumulate data to ascertain the principles of the change in the energy barrier to inversion of nitrogen in aziridine derivatives.

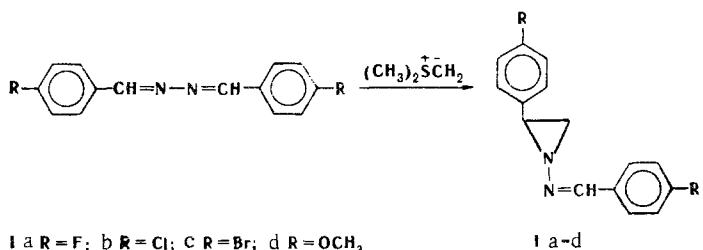
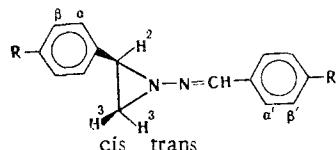


TABLE 1. Physicochemical Characteristics of Arylhydrazones Ia-d

| Com-<br>ound | mp, °C  | Empirical for-<br>mula   | Found, % |     |      | Calc., % |     |      | UV spectra*             |                        | Yield,<br>% |
|--------------|---------|--|----------|-----|------|----------|-----|------|-------------------------|------------------------|-------------|
|              |         |  | C        | H   | N    | C        | H   | N    | $\lambda_{max}$ ,<br>nm | $\epsilon \times 10^4$ |             |
| Ia           | 49–51   | C <sub>15</sub> H <sub>12</sub> F <sub>2</sub> N <sub>2</sub>  | 69.7     | 4.8 | 10.7 | 69.7     | 4.6 | 10.8 | 253                     | 0.74                   | 58          |
| Ib           | 72–73   | C <sub>15</sub> H <sub>12</sub> Cl <sub>2</sub> N <sub>2</sub> | 60.8     | 4.5 | 9.4  | 61.1     | 4.1 | 9.5  | 262                     | 0.66                   | 54          |
| Ic           | 115–116 | C <sub>15</sub> H <sub>12</sub> Br <sub>2</sub> N <sub>2</sub> | 47.2     | 3.1 | 7.2  | 47.4     | 3.1 | 7.3  | 265                     | 0.60                   | 61          |
| Id           | 82–83   | C <sub>17</sub> H <sub>18</sub> O <sub>2</sub> N <sub>2</sub>  | 72.3     | 6.4 | 10.0 | 72.3     | 6.4 | 9.9  | 272                     | 0.68                   | 64          |

\*The electronic spectra of Ia-d were obtained from n-hexane solutions.

TABLE 2. Parameters of the PMR Spectra of Hydrazones of the Ia-d Type



| Com-<br>ound | Chemical shifts, $\tau$ , ppm, relative to tetramethylsilane |                       |                         |      |          |         |           |          | SSCC*, Hz                   |                   |                 |
|--------------|--|-----------------------|-------------------------|------|----------|---------|-----------|----------|-----------------------------|-------------------|-----------------|
|              | H <sup>1</sup>   | H <sup>2</sup><br>cis | H <sup>2</sup><br>trans | -CH= | $\alpha$ | $\beta$ | $\alpha'$ | $\beta'$ | R                           | $H^1H^1$<br>trans | $H^1H^2$<br>cis |
| Ia           | 7.00   | 7.77                  | 7.57                    | 1.57 | 2.80     | 3.05    | 2.41      | 3.00     | —                           | 4.7               | 7.6             |
| Ib           | 7.02   | 7.78                  | 7.57                    | 1.67 | 2.89     | 2.89    | 2.53      | 2.77     | —                           | 4.7               | 7.8             |
| Ic           | 7.02   | 7.77                  | 7.57                    | 1.67 | 2.97     | 2.67    | 2.70      | 2.70     | —                           | 4.8               | 7.7             |
| Id           | 7.06   | 7.81                  | 7.62                    | 1.62 | 2.90     | 3.29    | 2.51      | 3.24     | 6.36<br>(R')<br>6.39<br>(R) | 4.9               | 7.7             |

\*The abbreviation SSCC represents spin-spin coupling constant;  
 $^2J = 0$ .

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The physicochemical characteristics of the synthesized compounds (Ia-d) are presented in Table 1, and the parameters of the PMR spectra are given in Table 2.

LITERATURE CITED

1. S. A. Giller, A. V. Eremeev, M. Yu. Lidak, V. A. Pestunovich, É. É. Liepin'sh, and I. Ya. Kalvin'sh, Khim. Geterotsikl. Soedin., 45 (1971).